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Hydrogen Isotopes

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Print Publication Year: 2007 Published Online: 2009

Publisher: Oxford University Press

ISBN: eISBN:

DOI: 10.1093/jama/9780195176339.022.503

Item type: section

Two isotopes of hydrogen have their own specific names, deuterium and tritium, which should be used instead of “hydrogen 2” and “hydrogen 3.” In text, the specific names are also preferred to the symbols 2H or D (for deuterium, which is stable) and 3H (for tritium, which is radioactive). The 2 forms of heavy water, D_2O and $3\text{H}_2\text{O}$, should be referred to by the approved nonproprietary names deuterium oxide and tritiated water, respectively. |